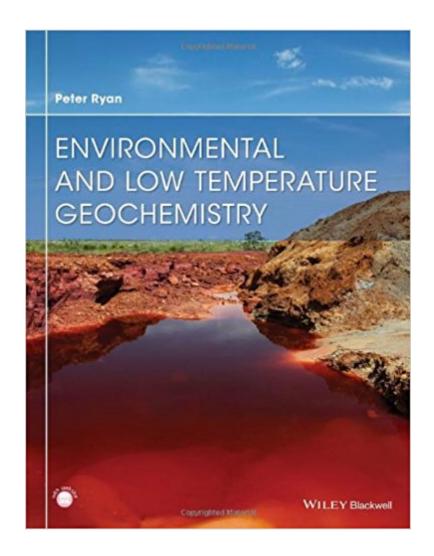


The book was found

Environmental And Low Temperature Geochemistry





Synopsis

Environmental and Low-Temperature Geochemistry presents conceptual and quantitative principles of geochemistry in order to foster understanding of natural processes at and near the earthâ ™s surface, as well as anthropogenic impacts on the natural environment. Â It provides the reader with the essentials of concentration, speciation and reactivity of elements in soils, waters, sediments and air, drawing attention to both thermodynamic and kinetic controls. A Specific features include: â ¢ An introductory chapter that reviews basic chemical principles applied to environmental and low-temperature geochemistry â ¢ Explanation and analysis of the importance of minerals in the environment â ¢ Principles of aqueous geochemistry â ¢ Organic compounds in the environment â ¢ The role of microbes in processes such as biomineralization, elemental speciation and reduction-oxidation reactions â ¢ Thorough coverage of the fundamentals of important geochemical cycles (C, N, P, S) â ¢ Atmospheric chemistry â ¢ Soil geochemistry â ¢ The roles of stable isotopes in environmental analysis â ¢ Radioactive and radiogenic isotopes as environmental tracers and environmental contaminants â ¢ Principles and examples of instrumental analysis in environmental geochemistry The text concludes with a case study of surface water and groundwater contamination that includes interactions and reactions of naturally-derived inorganic substances and introduced organic compounds (fuels and solvents), Â and illustrates the importance of interdisciplinary analysis in environmental geochemistry. Readership: Advanced undergraduate and graduate students studying environmental/low T geochemistry as part of an earth science, environmental science or related program. Additional resources for this book can be found at: www.wiley.com/go/ryan/geochemistry.

Book Information

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Customer Reviews

â œA useful acquisition for libraries serving undergraduate and graduate earth science programs. Summing Up: Recommended. Upper-division undergraduates, graduate students, and faculty.â •Â (Choice, 1 December 2014) Â

Environmental and Low-Temperature Geochemistry presents conceptual and quantitative principles of geochemistry in order to foster understanding of natural processes at and near the earthâ ™s surface, as well as anthropogenic impacts on the natural environment. Â It provides the reader with the essentials of concentration, speciation and reactivity of elements in soils, waters, sediments and air, drawing attention to both thermodynamic and kinetic controls. A Specific features include: â ¢ An introductory chapter that reviews basic chemical principles applied to environmental and low-temperature geochemistry â ¢ Explanation and analysis of the importance of minerals in the environment â ¢ Principles of aqueous geochemistry â ¢ Organic compounds in the environment â ¢ The role of microbes in processes such as biomineralization, elemental speciation and reduction-oxidation reactions â ¢ Thorough coverage of the fundamentals of important geochemical cycles (C, N, P, S) â ¢ Atmospheric chemistry â ¢ Soil geochemistry â ¢ The roles of stable isotopes in environmental analysis â ¢ Radioactive and radiogenic isotopes as environmental tracers and environmental contaminants â ¢ Principles and examples of instrumental analysis in environmental geochemistry The text concludes with a case study of surface water and groundwater contamination that includes interactions and reactions of naturally-derived inorganic substances and introduced organic compounds (fuels and solvents), Â and illustrates the importance of interdisciplinary analysis in environmental geochemistry. Readership: Advanced undergraduate and graduate students studying environmental/low T geochemistry as part of an earth science, environmental science or related program.

The book does a good job of presenting the material. However, there are so many typos that it's a little disconcerting.

Saved me in Environmental Geochemistry class. Thank you Peter Ryan!!!

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